

# Mass-Mass Stoichiometry Math for MIS

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→ only value to include 0  
in  $\mathbb{N}$ ? use  $\mathbb{Z}^+$ ?



ex Na, H, Cl

(1, Na), (1, H), (2, H)

## Types

ElementT =  $\{ \text{H, He, Li, Be, B, C, ...} \} \leftarrow$  not strings

MoleculeT = tuple of (num:  $\mathbb{N}$ , elem: ElementT)

CompoundT = Set of MoleculeT  $\leftarrow$  this says order doesn't matter

~~CompoundT = tuple of (num:  $\mathbb{N}$ , Molec: MoleculeT)~~  
~~tuple of (num:  $\mathbb{N}$ , Compound: CompoundT)~~

$\{ \text{Na} \}$   
 $\{ (1, \text{Na}), (1, \text{H}), (1, \text{Cl}) \}$   
 $\{ (1, \text{Na}), (1, \text{H}), (2, \text{H}) \}$

## etc

## Stoichiometry

StoichiometricT = tuple of (coeff:  $\mathbb{N}$ , Compound: CompoundT)

(2,  $\{ (1, \text{Na}) \}$ )  
(2,  $\{ (1, \text{H}), (1, \text{Cl}) \}$ )  
(2,  $\{ (1, \text{Na}), (1, \text{H}), (1, \text{Cl}) \}$ )  
(1,  $\{ (2, \text{H}) \}$ )

Chemical EqT = set of Compound StoichiometricT

ReactionT = sequence [2] of Chemical EqT

## Functions

atomicMass: ElementT  $\rightarrow \mathbb{R}$

atomicMass(e)  $\equiv (e = \text{H} \Rightarrow 1.0079 \mid e = \text{He} \Rightarrow 4.0026 \mid \dots)$

numAtomsInMolec: MoleculeT  $\times$  ElementT  $\rightarrow \mathbb{N}$

numAtomsInMolec( $m, e$ )  $\equiv (m.\text{elem} = e \Rightarrow m.\text{num} \mid m.\text{elem} \neq e \Rightarrow 0)$

numAtomsInCompound: CompoundT  $\times$  ElementT  $\rightarrow \mathbb{N}$

numAtomsInCompound(C, e)  $\equiv + (m: \text{MoleculeT} \mid m \in C \cdot \text{numAtomsInMolec}(m, e))$

Gros & Schneider notation  $\rightarrow (*)(x: T \mid R \cdot E)$

numAtomsInStoichiometric: StoichiometricT  $\times$  ElementT  $\rightarrow \mathbb{N}$

numAtomsInStoichiometric(s, e)  $\equiv s.\text{coeff} * \text{numAtomsInCompound}(s.\text{Compound}, e)$